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## **Amendments to the Claims:**

This listing of claims will replace all prior versions, and listings, of claims in the application:

1. (currently amended) A method for forming a supported palladium membrane used for hydrogen purification and production, comprising steps of:

providing a porous stainless steel support;

mechanically polishing said porous stainless steel support by one of an abrasive paper and an ultrasonic vibration;

electro-polishing said porous stainless steel support;

acid-washing said porous stainless steel support with an acid solution;

activating said porous stainless steel support by heating;

filling said porous stainless steel support with a metal;

electroless plating a palladium membrane on said <u>porous stainless steel</u> support with a palladium salt solution; <del>and</del>

DC sputtering an additional palladium membrane further on said <u>porous stainless steel</u> support:

annealing said palladium membrane.

- 2. (original) The method according to claim 1 wherein said support is a porous stainless steel support.
- 3. (canceled).
- 4. (original) The method according to claim 1 wherein said metal is one selected from a group consisting of palladium, niobium, tantalum and a combination thereof.
- 5. (original) The method according to claim 1 wherein said metal is a hydrogen permeable fine metal powder.
- 6. (original) The method according to claim 5 wherein said metal powder is mixed with one of a palladium paste and a high temperature epoxy resin.
- 7. (canceled).

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- 8. (canceled).
- 9. (original) The method according to claim 1 wherein said palladium salt solution contains 4.2~5.4 g/L Pd(NH<sub>3</sub>)<sub>4</sub>Cl<sub>2</sub>, 60~74 g/L EDTA, 600~700 g/L NH<sub>4</sub>OH and 0.32~0.4 c.c./L NH<sub>2</sub>NH<sub>2</sub>.
- 10. (original) The method according to claim 1 wherein said electroless plating is performed for 120~360 minutes.
- 11. (original) The method according to claim 1 wherein said electroless plating is performed at 50~70 °C.
- 12. The method according to claim 1 wherein a target of said DC sputtering is (original) 99~99.9% palladium.
- 13. (original) The method according to claim 1 wherein said DC sputtering is performed under a vacuum pressure of  $10^{-2} \sim 10^{-5}$  torr and a power input of  $200 \sim 500$  W at  $25 \sim 250$  °C.
- 14. (original) The method according to claim 1 wherein said DC sputtering is performed for 60~120 minutes.
- 15. The method according to claim 1 wherein said palladium membrane has a (original) thickness of 3~30 µm after said DC sputtering.
- 16. (currently amended) The method according to claim 1 further comprising a step of wherein said annealing said palladium membrane step is performed at a temperature ranged from 450 to ~550 °C under a nitrogen atmosphere including 3~10% a hydrogen concentration ranged from 3% to 10% for a period ranged from 4 to ~8 hours.
- 17. (currently amended) A method for forming a supported palladium membrane used for hydrogen purification and production, comprising steps of:

providing a porous stainless steel support;

mechanically polishing said porous stainless steel support by one of an abrasive paper and an ultrasonic vibration;

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electro-polishing said porous stainless steel support;

acid-washing said porous stainless steel support with an acid solution;

activating said porous stainless steel support by heating;

filling said porous stainless steel support with a metal; and

electroless plating a palladium membrane on said <u>porous stainless steel</u> support with a palladium salt solution; <u>and</u>

annealing said palladium membrane.

18. (currently amended) A method for forming a supported Pd/Ag membrane used for hydrogen purification and production, comprising steps of:

providing a porous stainless steel support;

mechanically polishing said porous stainless steel support by one of an abrasive paper and an ultrasonic vibration;

electro-polishing said porous stainless steel support;

acid-washing said porous stainless steel support with an acid solution;

activating said porous stainless steel support by heating:

filling said porous stainless steel support with a metal;

electroless plating a palladium membrane on said <u>porous stainless steel</u> support with a palladium salt solution;

electroless plating a silver membrane on said <u>porous stainless steel</u> support with a silver salt solution;

annealing said palladium membrane and said silver membrane to form a Pd/Ag membrane; and

DC sputtering an additional Pd/Ag membrane further on said <u>porous stainless steel</u> support.

- 19. (original) The method according to claim 18 wherein said support is a porous stainless steel support.
- 20. (canceled).
- 21. (original) The method according to claim 18 wherein said metal is one selected from a group consisting of palladium, niobium, tantalum and a combination thereof.

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- 22. (original) The method according to claim 18 wherein said metal is a hydrogen permeable fine metal powder.
- 23. (original) The method according to claim 22 wherein said metal powder is mixed with one of a palladium paste and a high temperature epoxy resin.
- 24. (canceled).
- 25. (canceled).
- 26. (original) The method according to claim 18 wherein said palladium salt solution contains  $4.2\sim5.4$  g/L Pd(NH<sub>3</sub>)<sub>4</sub>Cl<sub>2</sub>,  $60\sim74$  g/L EDTA,  $600\sim700$  g/L NH<sub>4</sub>OH and  $0.32\sim0.4$  c.c./L NH<sub>2</sub>NH<sub>2</sub>.
- 27. (original) The method according to claim 18 wherein said silver salt solution contains  $2.1\sim1$  g/L AgNO<sub>3</sub>,  $60\sim74$  g/L EDTA,  $600\sim700$  g/L NH<sub>4</sub>OH and  $0.32\sim0.4$  c.c./L NH<sub>2</sub>NH<sub>2</sub>.
- 28. (original) The method according to claim 18 wherein said electroless plating is performed at  $50\sim70$  °C.
- 29. (original) The method according to claim 18 wherein a target of said DC sputtering is a Pd/Ag alloy with a weight composition ratio of 77/23~60/40.
- 30. (original) The method according to claim 18 wherein said DC sputtering is performed under a vacuum pressure of  $10^{-2} \sim 10^{-5}$  torr and a power input of  $200 \sim 500$  W at  $25 \sim 250$  °C.
- 31. (original) The method according to claim 18 wherein said step of annealing said palladium membrane and said silver membrane is performed at  $450\sim550$  °C under a nitrogen atmosphere including  $3\sim10\%$  hydrogen for  $4\sim8$  hours.
- 32. (original) The method according to claim 18 wherein said palladium membrane has a thickness of  $3\sim30~\mu m$  after said DC sputtering.

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33. (currently amended) A method for forming a supported Pd/Ag membrane used for hydrogen purification and production, comprising steps of:

providing a porous stainless steel support;

mechanically polishing said porous stainless steel support by one of an abrasive paper and an ultrasonic vibration;

electro-polishing said porous stainless steel support;

acid-washing said porous stainless steel support with an acid solution; and

activating said porous stainless steel support by heating;

filling said porous stainless steel support with a metal;

electroless plating a palladium membrane on said <u>porous stainless steel</u> support with a palladium salt solution;

electroless plating a silver membrane on said <u>porous stainless steel</u> support with a silver salt solution; and

annealing said palladium membrane and said silver membrane to form a Pd/Ag membrane.

- 34. (new) The method according to claim 1 wherein said acid solution is HCl and said concentration thereof is ranged from 8 to 10 M HCl.
- 35. (new) The method according to claim 1 wherein said activating step is performed under a temperature ranged from 40 to 60 °C.
- 36. (new) The method according to claim 17 wherein said acid solution is HCl and said concentration thereof is ranged from 8 to 10 M HCl.
- 37. (new) The method according to claim 17 wherein said activating step is performed under a temperature ranged from 40 to  $60\,^{\circ}$ C.
- 38. (new) The method according to claim 17 wherein said annealing step is performed at a temperature ranged from 450 to 550 °C under a nitrogen atmosphere including a hydrogen concentration ranged from 3 to 10% for a period ranged from 4 to 8 hours.
- 39. (new) The method according to claim 18 wherein said acid solution is HCl and said concentration thereof is ranged from 8 to 10 M HCl.

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40. (new) The method according to claim 18 wherein said activating step is performed under a temperature ranged from 40 to  $60\,^{\circ}$ C.

- 41. (new) The method according to claim 33 wherein said acid solution is HCl and said concentration thereof is ranged from 8 to 10 M HCl.
- 42. (new) The method according to claim 33 wherein said activating step is performed under a temperature ranged from 40 to 60 °C.
- 43. (new) The method according to claim 33 wherein said step of annealing said palladium membrane and said silver membrane is performed at a temperature ranged from 450 to 550 °C under a nitrogen atmosphere including a hydrogen concentration ranged from 3 to 10% for a period ranged from 4 to 8 hours.